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# Structure inversion asymmetry enhanced electronic structure and electrical transport in 2D $A_3\text{SnO}$ ( $A = \text{Ca}, \text{Sr}, \text{and Ba}$ ) anti-perovskite monolayers

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## ABSTRACT

Anti-perovskites  $A_3\text{SnO}$  ( $A = \text{Ca}, \text{Sr}, \text{and Ba}$ ) are an important class of materials due to the emergence of Dirac cones and tiny mass gaps in their band structures originating from an intricate interplay of crystal symmetry, spin–orbit coupling, and band overlap. This provides an exciting playground for modulating their electronic properties in the two-dimensional (2D) limit. Herein, we employ first-principles density functional theory (DFT) calculations by combining dispersion-corrected SCAN + rVV10 and mBJ functionals for a comprehensive side-by-side comparison of the structural, thermodynamic, dynamical, mechanical, electronic, and thermoelectric properties of bulk and monolayer (one unit cell thick)  $A_3\text{SnO}$  anti-perovskites. Our results show that 2D monolayers derived from bulk  $A_3\text{SnO}$  anti-perovskites are structurally and energetically stable. Moreover, Rashba-type splitting in the electronic structure of  $\text{Ca}_3\text{SnO}$  and  $\text{Sr}_3\text{SnO}$  monolayers is observed owing to strong spin–orbit coupling and inversion asymmetry. On the other hand, monolayer  $\text{Ba}_3\text{SnO}$  exhibits Dirac cone at the high-symmetry  $\Gamma$  point due to the domination of band overlap. Based on the predicted electronic transport properties, it is shown that inversion asymmetry plays an essential character such that the monolayers  $\text{Ca}_3\text{SnO}$  and  $\text{Sr}_3\text{SnO}$  outperform thermoelectric performance of their bulk counterparts.

## KEYWORDS

electrical transport, anti-perovskites, low-dimensional materials, electronic structure, mechanical properties

## 1 Introduction

Until the discovery of two-dimensional quantum Hall effect (2D-QHE) [1], all phase transitions in crystalline materials could be explained well through Landau's theory of spontaneous symmetry breaking [2]. Since the integer Hall conductance in 2D-QHE arose without breaking any crystal symmetry, a new field of study in condensed matter physics came into existence based on the works of Thouless et al. [3] and Haldane [4], whereby transition between electronic states could be explained on the basis of topological phase transitions [5]. These pioneering works are now known to apply to a large range of 2D and three-dimensional (3D) materials and provide an effective tool for explaining phenomena beyond Landau theory. In the 2D realm, for instance, the experimental realization of monolayer (ML) graphene [6] has paved the way for the rapid advancement in synthesis and understanding of 2D materials, which were formerly considered to be thermodynamically unstable [2, 7]. Moreover, graphene also enabled the very first room temperature measurement of QHE resistance [8], showing that experimental and theoretical research

efforts at the flat-land can be motivated for achieving diversification of physical properties in a 2D material that are not available in its 3D counterpart [9, 10]. As a result, we now have an extended family of 2D materials beyond elemental MLs [11–14].

In the backdrop of rapid discoveries in the 2D realm, little attention has unfortunately been paid to the 2D derivatives of perovskite-type materials, which are otherwise known to display a wide range of functional physical properties in their bulk form that are tunable to specific applications by manipulating their structure [15]. This can mainly be attributed to the fact that research efforts for 2D materials have been inspired by the mechanical exfoliation technique [16–18] and, therefore, exploring 2D layers of strongly bound bulk perovskites has remained a sort of “no-man's-land” since it is not possible to peel-off single layers. In addition, it was earlier believed that 2D limits of isotropically bonded perovskites are unachievable as a structural collapse below a critical film thickness of 5 unit cells (UC) would be triggered [19]. However, Ji et al. [20] successfully demonstrated that freestanding and structurally stable 2D  $\text{SrTiO}_3$  and  $\text{BiFeO}_3$

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perovskites with high crystalline quality can be synthesized by adopting the synthesis technique developed by Lu et al. [21]. This layer-by-layer grow-and-release bottom-up technique has not only allowed the successful synthesis of other technologically important 2D perovskite oxides [22, 23], it has also encouraged theoreticians to employ first-principles calculations for exploring advanced structure–property relationships [24–27] that were previously untapped due to non-suitability of other top-down techniques for obtaining 1 UC thick 2D MLs of materials adopting the perovskite structure [20].

Apart from their technological importance for electronic and optical devices, 2D materials have also been extensively explored for applications in energy harvesting devices. A promising avenue for utilizing 2D materials in this context is their employment in thermoelectric (TE) devices which can help reduce the increasing global energy demands [28]. For example, graphene is one of the most thoroughly explored 2D materials due to its good mechanical strength, large surface area, and suitable electrical conductivity ( $\sigma$ ) [29]. However, the semi-metallic nature of graphene leads to mediocre TE efficiency, for which various nano-level strategies have been proposed [30–32]. Unfortunately, these strategies fall short of being adopted for commercial purposes that have led researchers to explore other 2D materials for TE devices. Learning from the case of graphene where the opening of the band gap allows improvement in Seebeck coefficient ( $S$ ) [33, 34], it is intuitive that oxide materials showing a semi-metallic nature, arising from a combination of crystal symmetry, spin–orbit coupling (SOC), and band overlap, in their bulk may lose these characteristics at the 2D limit. In particular, the anti-perovskite  $A_3SnO$  ( $A = Ca, Sr, \text{ and } Ba$ ) wherein the anions ( $Sn^{2+}$  and  $O^{2-}$ ) and the cation ( $A^{2+}$ ) occupy atomic sites inverse to those of a normal perovskite, are a unique class of solid materials that exhibit Dirac nodes in their electronic structure and are therefore good candidates for technologies requiring a vanishing band gap.

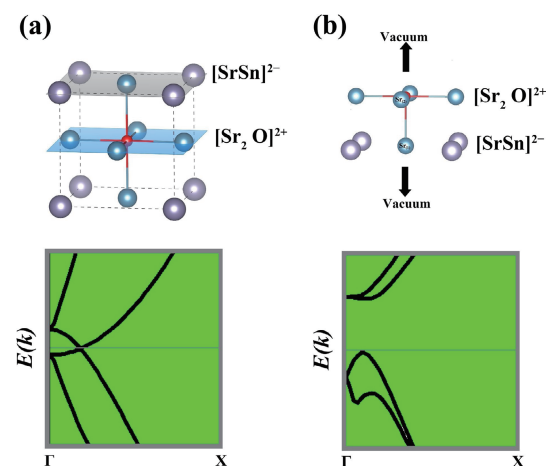
In the recent past, most of the theoretical and experimental research interest in anti-perovskites  $A_3SnO$  has mainly been motivated by trying to find an explanation behind the emergence of Dirac nodes in their band structure [35–45]. This has also led to the exploration of a wide array of structure–property relationships achievable in these materials, which give them flexible electronic, superconducting, and magnetic properties [46, 47]. Importantly, moderate values of  $S$ , low thermal conductivity ( $\kappa$ ), and metallic nature of resistivity observed in experiments [43, 44] have also encouraged theoretical inquiry into their TE response [45]. While both the bulk and surface electronic states of normal  $ABO_3$  perovskites show an insulating behavior [48, 49], the atomic arrangement of  $A_3SnO$  anti-perovskites can give rise to surface fermions that are attributed to the presence of inversion symmetry [39, 50–54]. The repulsion of conduction and valence band states for an electron precise  $A_3BO$  anti-perovskite [55] conventionally requires that a band gap should always be present near the Fermi level [56, 57]. However, because of an intricate interplay between crystal symmetry, SOC, and band inversion in the  $A_3BO$  compounds, an anti-crossing feature near the  $\Gamma$  symmetry point can be realized [40]. A vital feature arising out of the bands along the  $\Gamma$ –X path of an  $A_3BO$  anti-perovskite is the existence of symmetry protected Dirac nodes (Fig. 1(a)), which are influenced by SOC interaction and the degree of band inversion in A atom's d orbitals and Sn atom's p orbitals [52]. For instance, Ca and Sr containing anti-perovskites of Sn are trivial insulators and non-trivial topological crystalline insulators (TCIs), respectively [46, 58, 59]. On the other hand,  $Ba_3SnO$  (BSO) is predicted to be a topological semimetal exhibiting twin Dirac cones near the Fermi level [40].

Recent experiments employing molecular beam epitaxy (MBE)

and pulsed laser deposition (PLD) techniques for the synthesis of  $A_3SnO$  thin films on various substrates indicate that (001)-oriented surfaces are experimentally achieved showing metallic character [44, 58, 60, 61]. These results also agree well with density functional theory (DFT) based first-principles calculation for the (001) surface [62–64]. Interestingly, it has become evident that manipulating the inversion symmetry in  $A_3SnO$  can significantly impact the evolution of the surface electronic structure [64]. Since  $A_3SnO$  surfaces in thin films are constructed by stacking  $[A_2O]^{2+}$  and  $[ASn]^{2-}$  layers along [001] direction, a net dipole moment perpendicular to the (001) would bring about an electronic reconstruction that gives rise to metallic states [62–64]. For the case of symmetric surface terminations (i.e.,  $A_2O$ -terminated or  $ASn$ -terminated) studied for the bulk  $A_3SnO$ , electrons or holes from the bulk-like middle layer are available for electronic reconstruction to compensate for the polar catastrophe [64–66]. However, in the case of a 1 UC thick ML of  $A_3SnO$ , the absence of any bulk-like layer would result in a net electrostatic potential uncompensated by any electronic reconstructions when the band overlap is small. This scenario is depicted in Fig. 1, where one can see that for an asymmetric 2D system derived from the cubic space group #221 ( $Pm\bar{3}m$ ) of bulk UC of  $Sr_3SnO$  anti-perovskite consisting of one  $Sr_2O$  and one  $SrSn$  termination (Fig. 1(b)), the resulting space group #99 ( $P4mm$ ) lacks mirror symmetry as well as spatial inversion symmetry along the [001] direction. On the contrary, for a symmetric  $A_2O$ -terminated slab of anti-perovskites  $A_3SnO$  [62–64], both the inversion and mirror symmetries along [001] crystallographic direction are intact for the space group #123 ( $P4/mmm$ ). Therefore, an opening of the band gap at the 2D limit may only be realized for the asymmetric MLs of  $A_3SnO$ . Notably, lifting the spatial inversion symmetry would also cause spin degeneracy in the electronic states to be lifted by SOC, giving rise to the emergence of Rashba-type splitting [64, 67, 68]. Since spin-dependent band splitting introduced by the Rashba-type splitting is directly associated with improvement in  $S$ , the opening of band gap in 2D MLs of  $A_3SnO$  suggests that better TE performance can be achieved in these systems [69].

## 2 Computational details

The dispersion-corrected DFT calculations in the present work are used for investigating the stability, electronic structure, and thermoelectric properties of  $A_3SnO$ -MLs ( $A = Ca, Sr, \text{ and } Ba$ ). If not otherwise noted, the calculations for structural,



**Figure 1** Schematic illustration of the band topology for the case of (a) bulk UC of  $Sr_3SnO$  and (b) 2D  $Sr_3SnO$ -ML along the  $\Gamma$ –X path. The cubic bulk UC of  $Sr_3SnO$  in (a) is constructed by stacking  $[Sr_2O]^{2+}$  layer on top of a  $[SrSn]^{2-}$  layer along all simply crystallographic directions. On the other hand, the 2D  $Sr_3SnO$ -ML is constructed by stacking one  $[Sr_2O]^{2+}$  layer on top of one  $[SrSn]^{2-}$  layer along the [001] direction followed by vacuum region. The Sr, Sn, and O atoms are represented by turquoise, gray, and red sphere.

thermodynamic, electronic properties, and transport properties of all bulk UC and 1 UC thick ML are performed using the all-electronic full-potential linear-augmented plane wave (FP-LAPW) method available in the WIEN2k code [70]. We have employed the BoltzTrap2 package [71] for computing the thermoelectric properties. On the other hand, the Vienna *ab-initio* simulation package [72] in combination with the Phonopy code [73] is employed to examine the dynamical stability. For the WIEN2k based calculations, we have set the muffin-tin radii of Ca, Sr, Ba, Sn, and O to 2.0, 2.2, 2.25, 2.3, and 1.8, respectively, and the plane wave cut-off is set to  $RK_{\max} = 8.0$ . All the calculations are performed using an energy cut-off of  $-8.0$  Ry to separate core and valence states, while a large value of  $G_{\max}$  ( $24 \text{ Bohr}^{-1}$ ) is used. The bulk  $A_3\text{SnO}$  ( $A = \text{Ca, Sr, and Ba}$ ) is first optimized for obtaining the ground state lattice parameters using an energy convergence criteria of  $10^{-5}$  Ry. The ML systems are then derived from the bulk UC by introducing a vacuum region  $> 15 \text{ \AA}$ . These structural models are subsequently fully optimized using energy and force convergence criteria of  $10^{-5}$  Ry and  $1 \text{ mRy/a.u.}$ , respectively. For the bulk and 2D structural models of  $A_3\text{SnO}$ , we have employed  $15 \times 15 \times 15$  and  $15 \times 15 \times 1$   $k$ -meshes, respectively. On the other hand, the electronic density of states (DOS) and transport properties are computed using  $45 \times 45 \times 45$  and  $45 \times 45 \times 1$   $k$ -meshes, respectively. We note here that earlier DFT-based studies for the bulk UC of  $A_3\text{SnO}$  compounds report that a very dense  $k$ -mesh is necessary for obtaining a good approximation of the electronic density of states [45, 35]. However, despite the total energy convergence obtained with a shifted (a.k.a. “special”  $k$ -mesh)  $15 \times 15 \times 15$   $k$ -mesh, the complex electronic structures of bulk  $A_3\text{SnO}$  compounds are not fully accessible. On the contrary, an unshifted  $15 \times 15 \times 15$   $k$ -mesh is found to yield results in good agreement with a dense shifted or unshifted  $45 \times 45 \times 45$   $k$ -mesh (Fig. S1 in the Electronic Supplementary Material (ESM)). For this reason, in this study, we only present the electronic and transport properties computed using the unshifted  $k$ -mesh.

Since our calculations involve both bulk and 2D ML systems, the choice of DFT functional for a comparative analysis of the structural, thermodynamic, mechanical, and dynamical properties is crucial. While both generalized gradient approximation (GGA) and meta-GGA functionals are known to provide a good approximation of the physical properties of the bulk crystal, the inclusion of dispersion correction in these functionals does not necessarily work well for both worlds. In this context, the meta-GGA based SCAN + rVV10 functional is an exception that can simultaneously achieve a reliable account of the physical properties of both 3D and 2D realms [74, 75]. Our calculated results clearly show that inclusion of the dispersion correction through the rVV10 van der Waals (vdW) density functional [76] yields superior performance for structural and energetic properties of bulk  $A_3\text{SnO}$  compounds compared not only to previous calculations performed using the GGA-Perdew–Burke–Ernzerhof (PBE) functional [77] but also the bare SCAN meta-GGA. The SOC interactions in the  $A_3\text{SnO}$  systems are included through the second variational method [70] as these are necessary to capture the correct electronic structure of these materials. It is also worth mentioning here that recent DFT studies have demonstrated that SOC based electronic properties of  $A_3\text{SnO}$  compounds computed using the non-local modified Becke–Johnson (mBJ) potential functional [78, 79] are in better accord with experimental data compared to other approximations for the exchange-correlation potentials [45, 80]. Since a local modification of the mBJ potential functional designed for reproducing correct electronic properties of 2D materials better than the hybrid-DFT functionals has also recently been proposed [81, 82], we evaluate the electronic DOS, band structures, and transport properties of bulk UC and ML of

$A_3\text{SnO}$  systems using the semi-local and local modifications, respectively, of the mBJ potential functional.

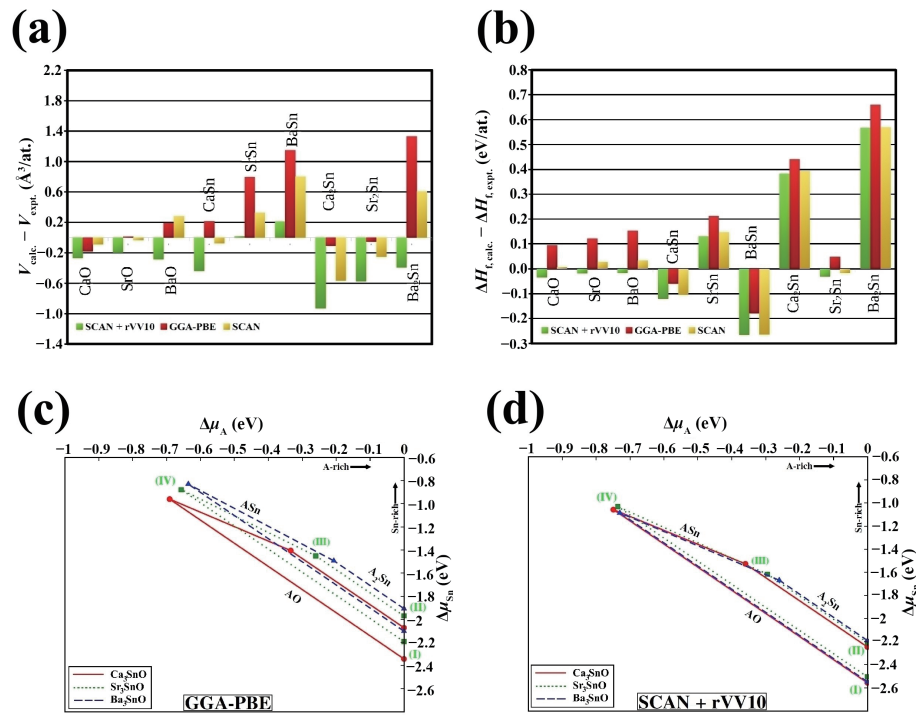
### 3 Results and discussion

#### 3.1 Structural, energetic, and thermodynamic properties of bulk $A_3\text{SnO}$

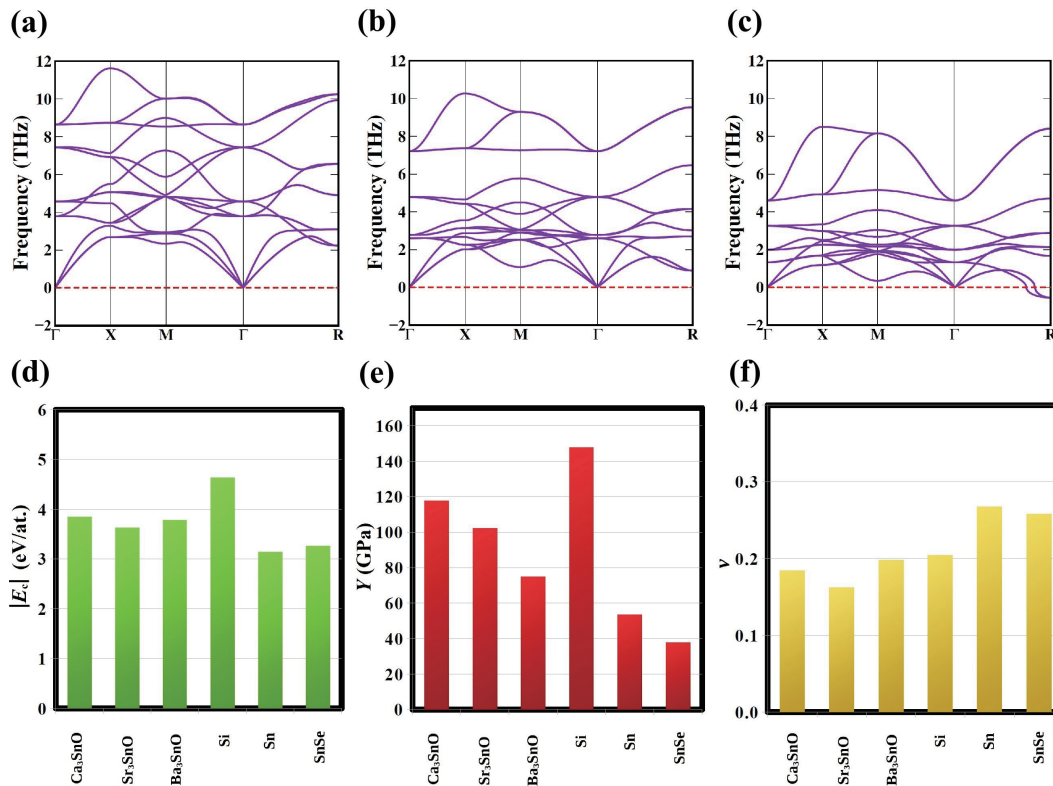
For a reliable account of the physical properties of bulk and ML modifications of  $A_3\text{SnO}$  systems, we begin our analysis by first evaluating the performance of SCAN + rVV10 exchange-correlation functionals of DFT in reproducing the structural and thermodynamic properties of the bulk systems. Although  $A_3\text{SnO}$  were first synthesized in the [83], little experimental data is available concerning the thermodynamic, dynamical, and mechanical stability of these compounds. However, only a couple of recent theoretical studies at the GGA-PBE level of DFT provide information about the thermodynamic stability of bulk  $A_3\text{SnO}$  compounds [64, 84, 85]. The better performance of the dispersion-corrected calculations in the present work compared to GGA-PBE and bare SCAN meta-GGA is evident from Figs. 2(a) and 2(b), where the reproducibility of the experimental data for the binary phases CaO, SrO, BaO, CaSn, SrSn, BaSn,  $\text{Ca}_2\text{Sn}$ ,  $\text{Sr}_2\text{Sn}$ , and  $\text{Ba}_2\text{Sn}$  (Tables S1 and S2 in the ESM) computed using SCAN + rVV10 show smaller average absolute deviation in the ground state volume ( $0.237 \text{ \AA}^3/\text{at.}$ ) and formation enthalpies ( $0.196 \text{ eV/at.}$ ). The better performance of SCAN + rVV10 approximation is further reflected in the good agreement of the calculated structural properties and the formation energies of bulk  $A_3\text{SnO}$  compounds (see Table S3 in the ESM). Moreover, the thermodynamic stability diagrams (details of the method adopted for computing stability diagrams are provided in Section S2 and S3 in the ESM) for the three  $A_3\text{SnO}$  compounds computed using SCAN + rVV10 (Fig. 2(d)) surprisingly show an almost overlapping stability region where similar values of the valid limits of the atomic chemical potentials of the Sn atom are obtained for the stable synthesis of bulk  $A_3\text{SnO}$  compounds. On the contrary, the atomic chemical potentials of Sn atom predicted by GGA-PBE (Fig. 2(c)) show a somewhat more dispersed range of the chemical potentials for these three compounds. Interestingly, the proximity of the stability line (I)–(IV) computed using SCAN + rVV10 functional for the three bulk  $A_3\text{SnO}$  compounds (restricting the formation of CaO, SrO, and BaO binary competing phases) is in excellent accord with experimental observation where better crystal quality as well as preclusion of AO phases is reported to be avoidable by a slightly excessive amount of alkaline-earth metal compared to what is stoichiometrically required for stable synthesis of these anti-perovskites [46].

#### 3.2 Dynamical, and mechanical properties of bulk $A_3\text{SnO}$

Figures 3(a)–3(c) show the phonon band structures of the three bulk  $A_3\text{SnO}$  compounds computed with the SCAN + rVV10 functional where a  $2 \times 2 \times 2$  supercell is used. All three compounds are found to be dynamically stable in the cubic anti-perovskite structure as no imaginary phonon modes (indicated by negative frequencies in the phonon band structure) are observed at the  $\Gamma$  symmetry point. It is worth recalling here that  $\text{SrTiO}_3$ , a prototypical cubic perovskite oxide at room temperature, is known to undergo an anti-ferrodistortive phase transition at low temperature. From a theoretical stand point, this low temperature transition of  $\text{SrTiO}_3$  is reflected in the imaginary phonon modes at the R symmetry point within the strongly-constrained and appropriately-normed (SCAN) based DFT calculations [86].



**Figure 2** Deviation of the SCAN + rVV10, GGA-PBE, and bare SCAN meta-GGA functionals' (a) ground state volume per atom and (b) formation enthalpies per atom from the experimental data of CaO, SrO, BaO, CaSn, SrSn, BaSn, Ca<sub>2</sub>Sn, Sr<sub>2</sub>Sn, and Ba<sub>2</sub>Sn binary compounds. (c) and (d) Show the thermodynamic stability regions of bulk UC of A<sub>3</sub>SnO (A = Ca, Sr, and Ba) compounds computed using GGA-PBE data reported in Ref. [77] and SCAN + rVV10 in this work, respectively.



**Figure 3** The phonon band structures of bulk UCs of (a) Ca<sub>3</sub>SnO, (b) Sr<sub>3</sub>SnO, and (c) Ba<sub>3</sub>SnO compounds computed using SCAN + rVV10 parameterization scheme. In ((d)–(f)) we present the absolute value of  $|E_c|$  (eV/at.),  $Y$  (GPa), and  $\nu$ , computed using SCAN + rVV10. For bulk diamond UCs of Si and Sn, and orthorhombic UC of SnSe, the experimental [89] cohesive energies and GGA-PBE [90] mechanical properties are plotted for comparison.

Nevertheless, cubic perovskite structure of SrTiO<sub>3</sub> is dynamically stable under ambient conditions. Since a recent low temperature experimental study has shown that bulk Ba<sub>3</sub>SnO undergoes a structural phase transition from the ideal cubic structure to an orthorhombic phase below 150 K [46], the negative phonon frequencies at the R symmetry point for bulk UC of Ba<sub>3</sub>SnO in Fig. 3(c) are indicative of the reliable performance of SCAN +

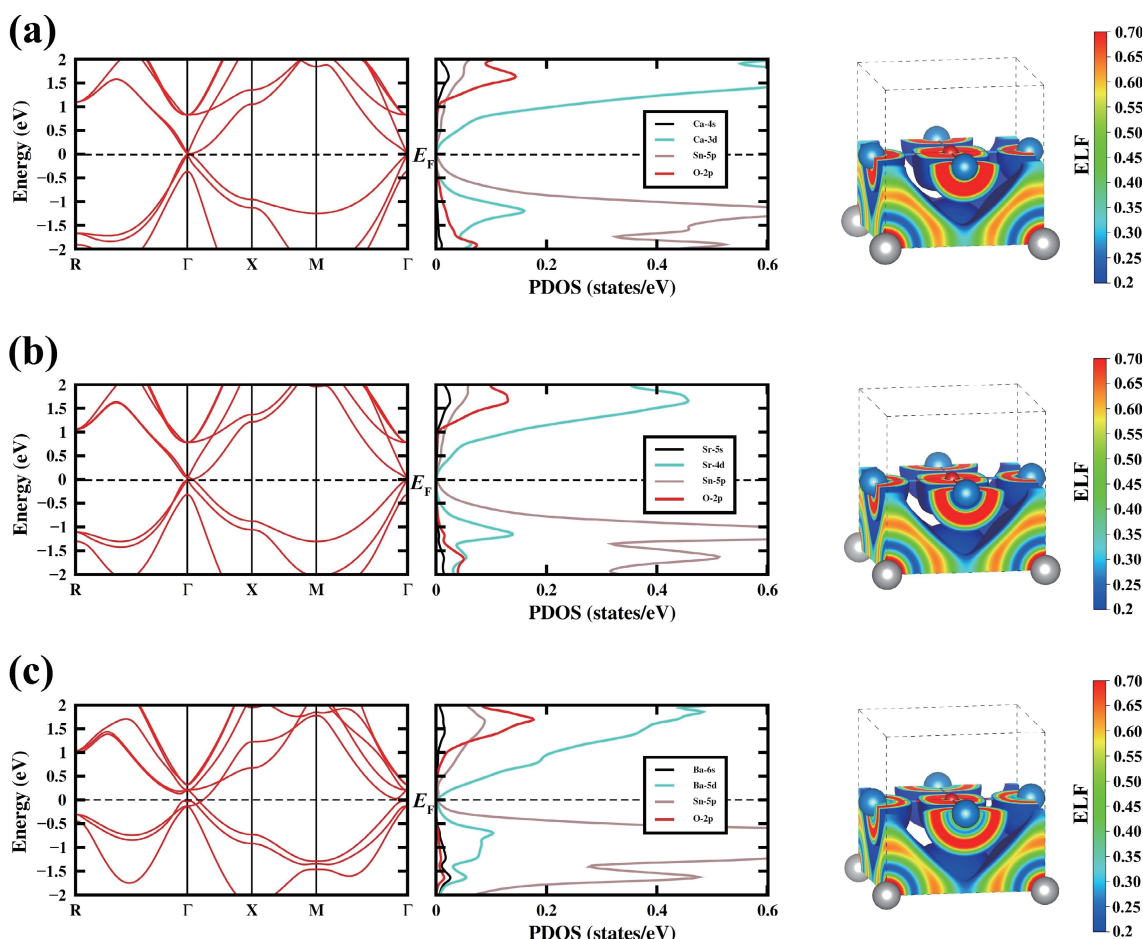
rVV10 calculations in predicting the dynamical properties of the three anti-perovskites studied in this work. This is further confirmed from the fact that both bulk UCs of Ca<sub>3</sub>SnO and Sr<sub>3</sub>SnO show no departure from the cubic *Pm-3m* phase at low temperatures [46].

Since cohesive energies and mechanical properties are commonly used for evaluating the stability of 2D materials [17],

for the sake of completeness we have also computed these values for the bulk UC of  $A_3\text{SnO}$  compounds. In Figs. 3(d)–3(f), we compare the cohesive energies ( $|E_c|$ ), Young's modulus ( $Y$ ), and Poisson's ratio ( $\nu$ ) of the bulk  $A_3\text{SnO}$  compounds with bulk UCs of Si, Sn, and SnSe that are also potential TE materials at their 2D limit [17, 87]. It is evident from Fig. 3(d) that all the bulk  $A_3\text{SnO}$  compounds have larger cohesive energies than the bulk UCs of Sn and SnSe. These findings, together with the large negative formation enthalpies for  $\text{Ca}_3\text{SnO}$  ( $-9.207$  eV/f.u.),  $\text{Sr}_3\text{SnO}$  ( $-8.676$  eV/f.u.), and  $\text{Ba}_3\text{SnO}$  ( $-8.259$  eV/f.u.) computed using the SCAN + rVV10 functional confirm the thermodynamic stability of these anti-perovskites. On the other hand, the SCAN + rVV10 calculated stiffness tensors (Table S4 in the ESM) confirm the mechanical stability (i.e.  $C_{11} - C_{12} > 0$ ,  $C_{11} - 2 \times C_{12} > 0$ ,  $C_{11} > 0$ , and  $C_{44} > 0$ ) of the bulk  $A_3\text{SnO}$  compounds [88]. Comparison of Young's modulus presented in Fig. 3(e) clearly shows that the bulk UCs of  $A_3\text{SnO}$  compounds are less rigid than Si. The SCAN + rVV10 predicted Poisson's ratio of all three  $A_3\text{SnO}$  compounds are found to be smaller than those reported for bulk UCs of Si, Sn, and SnSe; indicating that a smaller transverse strain corresponding to linear strain may be observed in these materials. It is worth pointing out here that GGA-PBE predicts an auxetic behavior for bulk UC of  $\text{Sr}_3\text{SnO}$  [90]. In our SCAN + rVV10 calculations, on the other hand, no such behavior is seen for the three bulk UCs of  $A_3\text{SnO}$  compounds. All in all, the structural, thermodynamic, dynamical, and mechanical properties computed using the SCAN + rVV10 functional clearly show that this dispersion-corrected exchange-correlation parameterization scheme enables a better comparison with experimental data for bulk UCs of  $A_3\text{SnO}$  compounds.

### 3.3 Electronic properties of bulk $A_3\text{SnO}$

The electronic properties for the bulk UCs of three  $A_3\text{SnO}$  compounds computed using the SCAN + rVV10 functional indicate an anti-crossing feature along the  $\Gamma$ -X direction for both  $\text{Ca}_3\text{SnO}$  and  $\text{Sr}_3\text{SnO}$ . On the other hand,  $\text{Ba}_3\text{SnO}$  is found to exhibit a semi-metallic band structure with twin Dirac cones below the Fermi level along the  $\Gamma$ -X direction (Fig. S2 in the ESM). The electronic bands near the Fermi level computed using SCAN + rVV10 for  $\text{Sr}_3\text{SnO}$  and  $\text{Ba}_3\text{SnO}$  are qualitatively similar to the ones reported in earlier studies employing HSE06 and mBJ functionals [38, 40, 45, 52, 53, 84]. Since the applications of both the HSE06 hybrid functional [38] and the mBJ potential functional [45, 80] open up a band gap by slightly lifting the conduction band upwards along the  $\Gamma$ -X direction in bulk  $\text{Ca}_3\text{SnO}$ , in the present work, we focus mainly on the electronic properties predicted by the mBJ potential functional. In fact, electronic properties computed using the mBJ potential functional within the FP-LAPW method have already been shown to provide an excellent agreement with the band structures computed using the tight-binding model for the Wannier functions of the orbitals contributing near the Fermi level of bulk  $A_3\text{SnO}$  compounds [45]. Figure 4 presents the electronic band structure diagrams, and partial DOS plots of the bulk modifications of the three  $A_3\text{SnO}$  ( $A = \text{Ca}, \text{Sr},$  and  $\text{Ba}$ ) compounds computed using the mBJ parameterization scheme where one can see that the overlap of Ba-5d states and Sn-5p states in  $\text{Ba}_3\text{SnO}$  is the maximum below and above the Fermi level. In a recent study, we have performed a thorough analysis of differences in the ionic binding and metallic



**Figure 4** Electronic band structures (left panel) and partial density of states plots (middle panel) for bulk UCs of (a)  $\text{Ca}_3\text{SnO}$ , (b)  $\text{Sr}_3\text{SnO}$ , and (c)  $\text{Ba}_3\text{SnO}$  anti-perovskites computed using the non-local mBJ potential functional. The computed ELF for the three compounds within the 0.2–0.7 range are also shown in the right panel where an isosurface value of 0.2 is used (indicated by the blue colored surface) to show the changes in the metallic bonding between two Sn atoms in the ASn plane on going from  $\text{Ca}_3\text{SnO}$  to  $\text{Ba}_3\text{SnO}$ .

bonding in  $\text{Ca}_3\text{SnO}$  and  $\text{Ba}_3\text{SnO}$  [85], where smaller metallic interaction among Sn atoms in the BaSn layer of  $\text{Ba}_3\text{SnO}$  is found to be linked to the larger band overlap in this material. However, in order to compare the changes in the metallic bonding for three MLs derived from the bulk UCs of  $\text{A}_3\text{SnO}$  anti-perovskites (discussed later), we use the electron localization function (ELF) method of Becke and Edgecombe [91] for the structures optimized using the SCAN + rVV10 functional in the present study. Figure 4 again confirms the decreasing metallic interaction evident from the changes in the blue isosurfaces (corresponding to an ELF = 0.2 [92]) between two Sn atoms in the ASn layer on going from  $\text{Ca}_3\text{SnO}$  to  $\text{Ca}_3\text{SnO}$ . On the other hand, ionic bonding [84, 85] between the A and O atoms in the  $\text{A}_2\text{O}$  layer is also confirmed. Since the partial DOS plots computed using the mBJ potential functional (Fig. 4) provide a reliable account of the competition between the SOC and band overlap in bulk UCs of  $\text{A}_3\text{SnO}$  compounds, it is evident that this functional provides an accurate description of the electronic properties of the bulk  $\text{A}_3\text{SnO}$  when combined with the structures computed using the dispersion-corrected SCAN + rVV10 functional [93].

### 3.4 Thermodynamic stability of $\text{A}_3\text{SnO}$ -ML

For the three  $\text{A}_3\text{SnO}$ -ML ( $A = \text{Ca}, \text{Sr}, \text{and Ba}$ ) systems studied in the present work, the phonon band structure for the fully optimized  $1 \times 1 \times 1$  vacuum supercell is shown in Figs. 5(a)–5(c), which have been computed using the SCAN + rVV10 functional. It is evident that all three anti-perovskite ML systems are dynamically stable and show no imaginary phonons modes along the X– $\Gamma$ –M  $k$ -path. However, we note that small negative phonon frequencies emerge near the  $\Gamma$  point for larger (e.g.,  $3 \times 3 \times 1$ ) supercells of  $\text{Sr}_3\text{SnO}$ -ML (see Fig. S3 in the ESM). Since we also found that the phonon band structure of a  $2 \times 2 \times 1$  supercell of  $\text{SrTiO}_3$ -ML calculated using SCAN + rVV10 exhibits imaginary phonon modes which disappear only for a  $1 \times 1 \times 1$  supercell, this

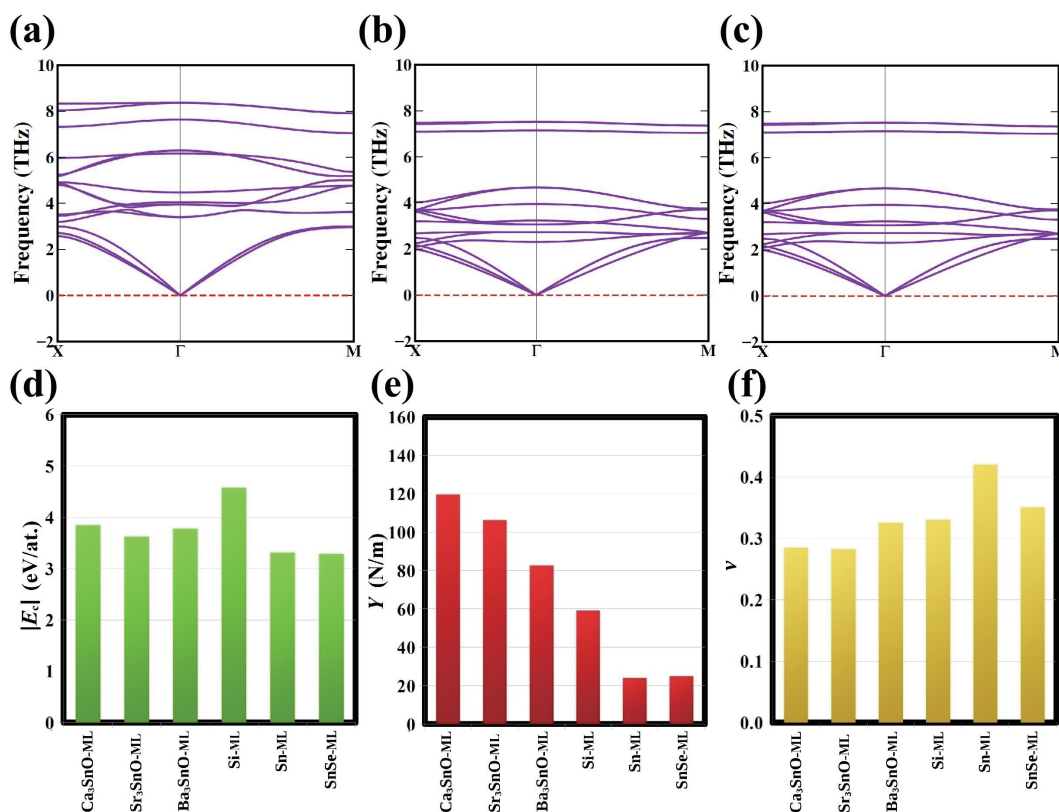
can be due to the presence of electrostatic potential between  $\text{A}_2\text{O}$  and ASn layers whose influence enhances with increasing size of the supercell and cannot fall to zero for an infinite vacuum region [25, 94]. Similar negative phonon modes are also reported in earlier DFT studies for large supercell calculations of perovskite-type MLs [25, 27, 16]. To this end, we resort to further examining the thermodynamic stability of the  $\text{A}_3\text{SnO}$ -MLs that can be assessed in terms of the surface energy,  $\gamma$ , which should be a small positive value [16, 95–97]. However, due to the asymmetric surface termination, one can compute an averaged surface energy,  $\hat{\gamma}$ , in terms of cleavage energy,  $\beta$ , which is defined as the energy required per unit area to create two free surfaces [97] and is given by the expression [96]

$$\beta = \frac{E_{\text{A}_3\text{SnO-ML}}^t - 3(E_A^t + \Delta\mu_A) - (E_{\text{Sn}}^t + \Delta\mu_{\text{Sn}}) - \frac{1}{2}(E_{\text{O}_2}^t + \Delta\mu_{\text{O}})}{A} \quad (1)$$

In Eq. (1),  $E_{\text{A}_3\text{SnO-ML}}^t$ ,  $E_A^t$ , and  $E_{\text{Sn}}^t$  and  $E_{\text{O}_2}^t$  are the ground state energies of  $\text{A}_3\text{SnO}$ -ML, face-centered cubic (fcc) UCs of  $A = \text{Ca}, \text{Sr}, \text{and Ba}$  and diamond UC of Sn and an  $\text{O}_2$  dimer computed using the SCAN + rVV10 functional. The atomic chemical potentials of A, Sn, and O in Eq. (1) are referenced to the thermodynamic stability diagram of bulk  $\text{A}_3\text{SnO}$  (Fig. 2(c)), while  $A$  is the surface area of the optimized  $\text{A}_3\text{SnO}$ -ML. It is worth noting here that writing cleavage energy in this way means that  $\beta = 2\hat{\gamma} = \gamma_1 + \gamma_2$ , such that for a 2D system with symmetric surface terminations  $\beta/2 = \hat{\gamma} = \gamma_1 = \gamma_2$ , while for an asymmetric 2D system such as  $\text{A}_3\text{SnO}$ -MLs studied here,  $\beta/2 = \hat{\gamma} \neq \gamma_1 \neq \gamma_2$ . Therefore, the averaged surface energy for the  $\text{A}_3\text{SnO}$ -MLs studied in this work can be written as

$$\hat{\gamma} = \frac{E_{\text{A}_3\text{SnO-ML}}^t - 3(E_A^t + \Delta\mu_A) - (E_{\text{Sn}}^t + \Delta\mu_{\text{Sn}}) - \frac{1}{2}(E_{\text{O}_2}^t + \Delta\mu_{\text{O}})}{2A} \quad (2)$$

The  $\hat{\gamma}$  values for  $\text{Ca}_3\text{SnO}$ -ML,  $\text{Sr}_3\text{SnO}$ -ML, and  $\text{Ba}_3\text{SnO}$ -ML are found to be 0.942, 0.751, and 0.551 J/m<sup>2</sup>, respectively, which are all



**Figure 5** The phonon band structures of (a)  $\text{Ca}_3\text{SnO}$ -ML, (b)  $\text{Sr}_3\text{SnO}$ -ML, and (c)  $\text{Ba}_3\text{SnO}$ -ML systems computed using SCAN + rVV10 parameterization scheme. In ((d)–(f)) we compare the absolute value of  $|E_c|$  (eV/at.),  $Y$  (N/m) and  $\nu$ , computed using SCAN + rVV10. For Si-ML (silicene), Sn-ML (stanene), and SnSe-ML the theoretical data corresponding to zig-zag direction is plotted for comparison [17].

smaller than  $\hat{\gamma} = 1.206 \text{ J/m}^2$  which we have computed as a benchmark in the present work using the same procedure for a fully optimized 1 UC thick SrTiO<sub>3</sub>-ML. Moreover,  $\hat{\gamma}$  of the studied ML systems are considerably smaller than surface energies of A<sub>2</sub>O-terminated and ASn-terminated (001) surfaces of bulk A<sub>3</sub>SnO [62, 63]. It is interesting to note that  $\hat{\gamma}$  for both asymmetric perovskite as well as anti-perovskite MLs decreases with increasing roughness of the surface (as discussed later). However, since all systems discussed here have a small positive value of  $\hat{\gamma}$ , it is clear that these 2D ML systems are thermodynamically stable.

### 3.5 Structural, energetic, and mechanical properties of A<sub>3</sub>SnO-ML

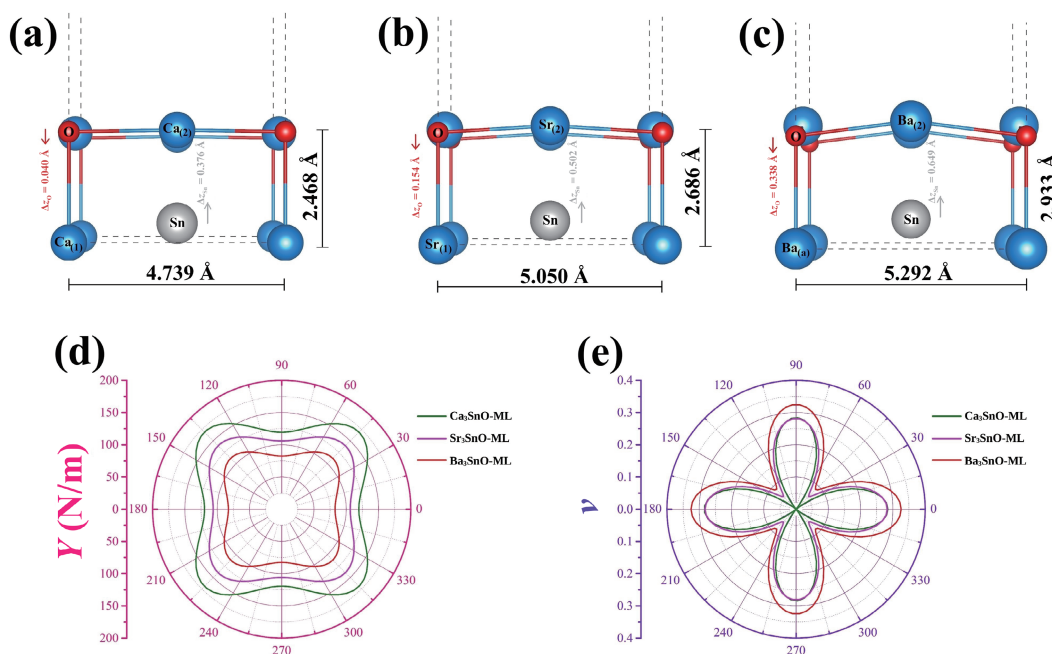
The stability of 2D A<sub>3</sub>SnO-ML systems is also confirmed from the cohesive energies presented in Fig. 5(d), which are larger than the theoretical values of  $|E_c|$  for stanene and SnSe-ML and comparable to the value of  $|E_c|$  predicted for silicene [17]. Using the SCAN + rVV10 functional, we have also computed the stiffness tensors  $C_{11}$ ,  $C_{22}$ ,  $C_{12}$ , and  $C_{66}$  by transforming the 3D units (N/m<sup>2</sup>) to 2D units (N/m) (Table S5 in the ESM) where the mechanical stability of the three A<sub>3</sub>SnO-MLs studied here is confirmed since all of  $C_{11} > 0$ ;  $C_{22} > 0$ ;  $C_{66} > 0$ ;  $C_{11} \times C_{22} - C_{12}^2 > 0$  are satisfied [88]. From Young's modulus presented in Fig. 5(e), it is evident that A<sub>3</sub>SnO-MLs are slightly more rigid than silicene, stanene, and SnSe-ML. However, the Poisson ratios of the A<sub>3</sub>SnO-ML are slightly enhanced compared to the bulk UCs of these systems, indicating that transverse strain corresponding to linear strain increases at the 2D limit of these anti-perovskites. However, the atomic structures of A<sub>3</sub>SnO-MLs are anisotropic since in addition to shrink in the in-plane lattice parameters, the atomic positions diverge from their ideal limits in the bulk UCs of A<sub>3</sub>SnO compounds. Figures 6(a)–6(c) show the in-plane  $a_0 = b_0$  lattice parameters of the three A<sub>3</sub>SnO-MLs, which have slightly shrunk compared to the lattice parameters of their corresponding bulk UCs (Table S3 in the ESM). Going from Ca<sub>3</sub>SnO-ML to Ba<sub>3</sub>SnO-ML, one can see that the upward shift of Sn atom from the ASn-layer and A<sub>(2)</sub> atom from the A<sub>2</sub>O-layer increases. This is also evident from the increasing values of the displacement of Sn (i.e.,  $\Delta z_{\text{Sn}}$ ) and O (i.e.,  $\Delta z_{\text{O}}$ ) atoms relative to the A<sub>(1)</sub> and A<sub>(2)</sub> atoms,

which result in the surface roughness. Such roughness has also been experimentally and theoretically observed in SrTiO<sub>3</sub>-ML, LaAlO<sub>3</sub>-ML, and BiFeO<sub>3</sub>-ML, which is necessary for stabilizing a freestanding 2D system [20, 25]. To examine the in-plane variation of Young's modulus and Poisson's ratio relative to the  $x$ -direction, we have plotted the two quantities as a function in-plane angle  $\theta$  in Figs. 6(d) and 6(e), respectively [98]. It is evident from Fig. 6(d) that the variation in Young's modulus does not exceed an increase of 30% compared to its values along the  $x$ -direction. On the other hand, the Poisson's ratio almost decreases to  $\sim 0$  for Sr<sub>3</sub>SnO-ML ( $\nu(45^\circ) = 0.064$ ) and Ba<sub>3</sub>SnO-ML ( $\nu(45^\circ) = 0.095$ ) and even becomes negative for Ca<sub>3</sub>SnO-ML ( $\nu(45^\circ) = -0.002$ ), indicating that this 2D material might expand perpendicular to an in-plane stretch [17].

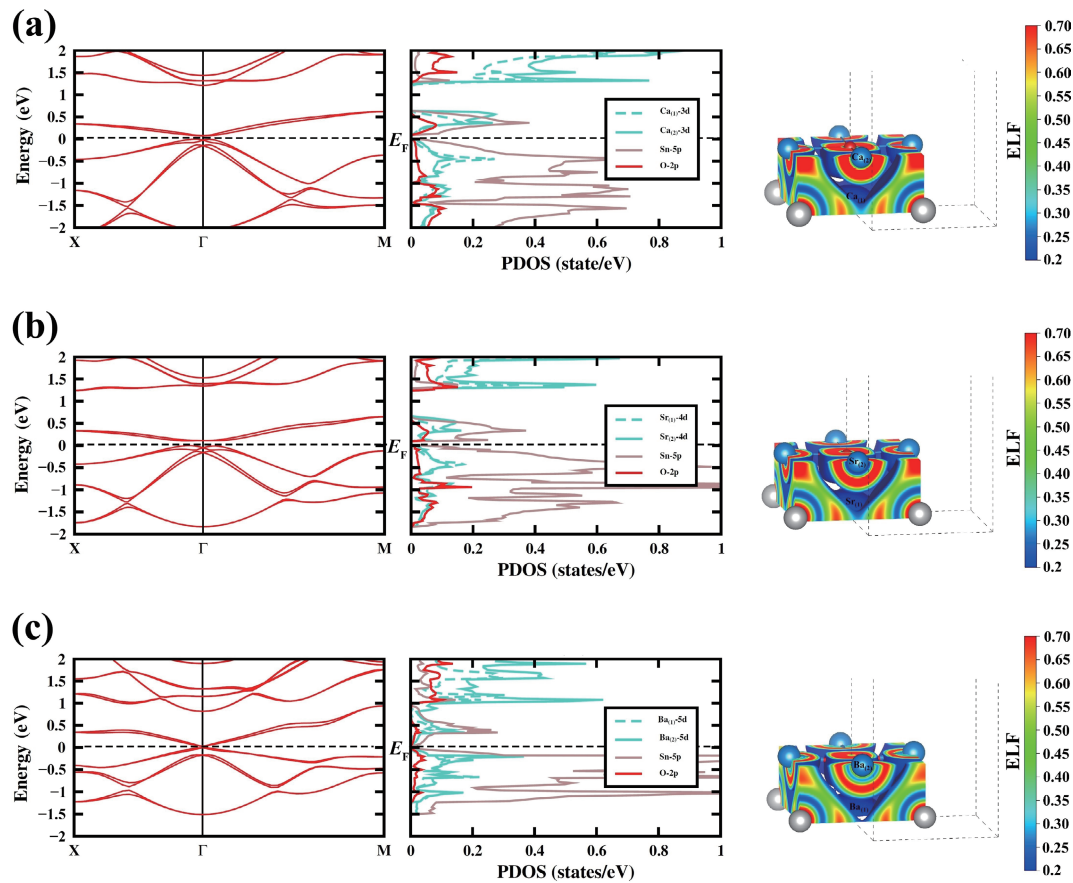
### 3.6 Electronic properties of A<sub>3</sub>SnO-ML

The electronic properties of the A<sub>3</sub>SnO-MLs computed using the local mBJ potential functional are presented in Fig. 7, where one can see that both Ca<sub>3</sub>SnO-ML and Sr<sub>3</sub>SnO-ML show an indirect band gap nature with small band gaps of 76 and 112 meV, respectively. The conduction band minima of both these ML systems are located at the  $\Gamma$  symmetry point, while the valence band maximum is found to be located slightly away from the  $\Gamma$  point owing to the emergence of Rashba-type splitting caused by the stronger influence of SOC interaction compared to the band overlap in these systems. On the other hand, the electronic band structure of Ba<sub>3</sub>SnO-ML does not show splitting in the valence bands, while Dirac cones appear at the  $\Gamma$  symmetry point with a tiny mass gap of 29 meV. Although the electronic properties computed using the SCAN + rVV10 functional show all three ML systems to have almost similar electronic structure (Fig. S4 in the ESM), one can again see the better performance of the mBJ potential functional in reproducing the competition between SOC and band overlap for the bulk UCs of A<sub>3</sub>SnO compounds reflected in the electronic structure of Ba<sub>3</sub>SnO-ML. Since bulk UC of Ba<sub>3</sub>SnO has a larger band overlap than Ca<sub>3</sub>SnO and Sr<sub>3</sub>SnO, a Rashba-type splitting in the valence band maximum is not seen in the electronic band structure of Ba<sub>3</sub>SnO-ML [67, 69, 99].

Similar to the case of A<sub>3</sub>SnO-ML studied in this work, Rashba-



**Figure 6** The fully optimized atomic structures of (a) Ca<sub>3</sub>SnO-ML, (b) Sr<sub>3</sub>SnO-ML and (c) Ba<sub>3</sub>SnO-ML systems having space group #99 (*P4mm*) showing the optimized in-plane lattice parameters and out-of-plane thickness along with the displacement of Sn (i.e.,  $\Delta z_{\text{Sn}}$ ) and O (i.e.,  $\Delta z_{\text{O}}$ ) atoms relative to the A<sub>(1)</sub> and A<sub>(2)</sub> atoms, respectively. (d) and (e) show Young's modulus and Poisson's ratio as a function of in-plane rotation angle  $\theta$ .



**Figure 7** Electronic band structures (left panel) and partial density of states plots (middle panel) for 1 ML (a)  $\text{Ca}_3\text{SnO}$ , (b)  $\text{Sr}_3\text{SnO}$ , and (c)  $\text{Ba}_3\text{SnO}$  anti-perovskites computed using the local version of mBJ parameterization scheme. The computed ELF for the three ML systems within the 0.2–0.7 range is shown in the right panel where an isosurface value of 0.2 is used (indicated by the blue surface coloring) to show the changes in the metallic bonding between two Sn atoms in the ASn layer on going from  $\text{Ca}_3\text{SnO}$ -ML to  $\text{Ba}_3\text{SnO}$ -ML. For comparison, we show the ELF plots for that section of the ML system's structural model adopting space group #99 ( $P4mm$ ) that can be directly compared with the ELF of bulk UC of  $\text{A}_3\text{SnO}$  anti-perovskite compounds displayed in Fig. 4.

type splitting in the electronic structure emerging due to the presence of inversion asymmetry has very recently been reported in experimental and theoretical studies carried out for other SOC dominated 2D perovskites [100, 101]. In order to examine the strength of the Rashba splitting evident in the electronic structures of  $\text{Ca}_3\text{SnO}$ -ML and  $\text{Sr}_3\text{SnO}$ -ML, we have computed the Rashba coefficient ( $\alpha_R$ ) using the expression

$$\alpha_R = 2E_R/k_R \quad (3)$$

where  $E_R$  is the energy difference between the band maximum and band degeneracy point (i.e.  $\Gamma$  symmetry point), and  $k_R$  is the momentum offset corresponding to these two energy values [102]. From the electronic band structures shown in Fig. 7, the Rashba coefficients for  $\text{Ca}_3\text{SnO}$ -ML and  $\text{Sr}_3\text{SnO}$ -ML are found to be 1.155 and 1.422 eV/Å, respectively. These values are clearly larger than other 2D systems [102] and comparable with the values reported for hybrid perovskites (1.6 eV/Å) [103] that are classified as conventional giant Rashba systems. It is worth mentioning here that the lifting of the band degeneracy caused by the asymmetric nature of the structural model or inclusion of SOC could result in the emergence of stable ferromagnetic (FM) ordering in  $\text{A}_3\text{SnO}$  systems [64, 84, 85]. This prospect is particularly important for the case of  $\text{Ca}_3\text{SnO}$ -ML and  $\text{Sr}_3\text{SnO}$ -ML systems studied in this work for which the emergence of Rashba-type splitting is caused by the stronger SOC and the presence of inversion asymmetry. By comparing the total energies of non-magnetic and FM calculations, we found an energy difference per surface area of  $< 0.1$  eV/Å<sup>2</sup> and a zero spin magnetic moment for the relaxed  $\text{A}_3\text{SnO}$ -ML systems. This is also confirmed from the spin texture associated (Fig. S5 in the ESM) with the Rashba-type splitting observed in the electronic band structure of  $\text{Sr}_3\text{SnO}$ -ML (Fig. 7(b))

where the iso-energy in-plane  $S_x$  and  $S_y$  components involved in the valence band splitting near the Fermi level have opposite spin directions, which is a typical Rashba pattern [102].

It is evident from the projected density of states (PDOS) plots shown in Fig. 7 that the  $\text{A}_{(1)}$  atom's d states contribute more in the valence band maximum than the  $\text{A}_{(2)}$  atom's d states for  $\text{Ca}_3\text{SnO}$ -ML and  $\text{Sr}_3\text{SnO}$ -ML, resulting in the Rashba-type splitting and small band gaps. On the other hand, the  $\text{Ba}_{(2)}$ -5d states show a slightly larger contribution than  $\text{Ba}_{(1)}$ -5d states in the valence band maximum of  $\text{Ba}_3\text{SnO}$ -ML. Moreover, the unoccupied mid-gap bands above the Fermi level have larger contributions from Sn-5p states in  $\text{Ca}_3\text{SnO}$ -ML and  $\text{Sr}_3\text{SnO}$ -ML compared to an almost equal contribution of  $\text{Ba}_{(2)}$ -5d and Sn-5p states above the Fermi level in  $\text{Ba}_3\text{SnO}$ -ML. The smaller overlap of the  $\text{A}_{(2)}$  atom's d states and Sn-5p causes the emergence of localized electrons at the  $\text{A}_2\text{O}$  surface of  $\text{Ca}_3\text{SnO}$ -ML and  $\text{Sr}_3\text{SnO}$ -ML, which is confirmed from the significant values of ELF ( $> 0.7$ ) for these systems compared to the metallic bonding ( $0.2 < \text{ELF} < 0.7$ ) for the in-plane ASn layer (Fig. S6 in the ESM). On the other hand, for the case of  $\text{Ba}_3\text{SnO}$ -ML, both in-plane and out-of-plane ELF remain in the range  $0.2 < \text{ELF} < 0.7$ . Since the conservation of electron localization function in the  $0.2 < \text{ELF} < 0.7$  range for  $\text{Ba}_3\text{SnO}$ -ML indicates significant charge transfer between the  $[\text{Ba}_2\text{O}]^{2+}$  and  $[\text{BaSn}]^{2-}$  layers, we quantitatively compare the charge localization at the  $\text{A}_2\text{O}$  surfaces by examining the changes in the effective Bader charges of the  $\text{A}_{(1)}$ ,  $\text{A}_{(2)}$ , Sn, and O atoms of the ML systems of  $\text{A}_3\text{SnO}$  in comparison to their bulk counterparts (Table S6 in the ESM). It is evident from the computed data that the effective Bader charges for  $\text{Ba}_{(1)}$ ,  $\text{Ba}_{(2)}$ , and Sn atoms  $\text{Ba}_3\text{SnO}$ -ML show larger deviation from their bulk values, indicating electronic reconstruction for this system

that conserves ELF in the  $0.2 < \text{ELF} < 0.7$  range. On the other hand, smaller deviations of the Bader charges for  $\text{Ca}_3\text{SnO-ML}$  and  $\text{Sr}_3\text{SnO-ML}$  confirm smaller charge transfer between the  $[\text{A}_2\text{O}]^{2+}$  and  $[\text{ASn}]^{2-}$  layers at the 2D limit such that localized electrons at the  $\text{A}_2\text{O}$  surface are responsible for the Rashba-type splitting near the valence band maximum which is a necessary ingredient for improving the electronic transport properties in these systems.

### 3.7 Transport properties

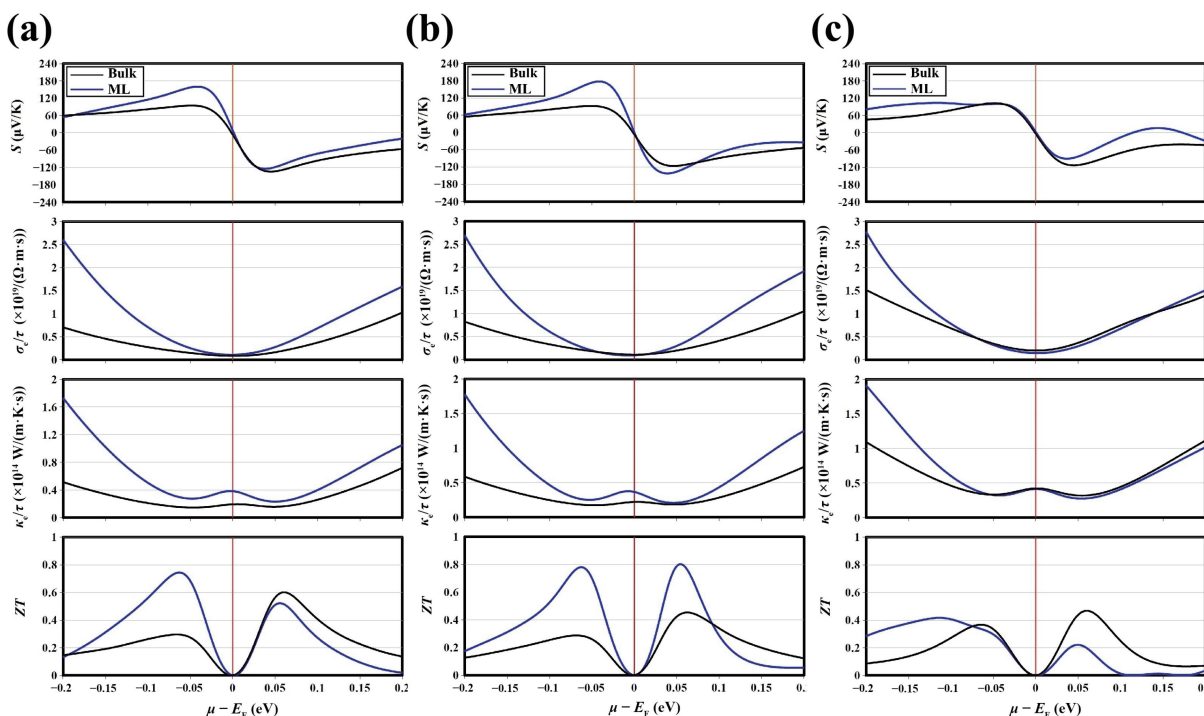
Comparing the electronic band structures of bulk UC and ML shown in Figs. 4 and 7, respectively, one can already expect that the electronic transport properties of  $\text{Ca}_3\text{SnO-ML}$  and  $\text{Sr}_3\text{SnO-ML}$  may have enhanced at the 2D limit. To examine this from an application point of view, in Fig. 8 we compare the  $S$ , electronic conductivity ( $\sigma_e/\tau$ ), and electronic thermal conductivity ( $\kappa_e/\tau$ ) of the bulk UCs and 2D MLs of  $\text{A}_3\text{SnO}$  as a function of chemical potential at 300 K. It is important to point out here that for the  $\sigma_e/\tau$  and  $\kappa_e/\tau$  values presented in Fig. 8, the shortcoming of the BoltzTrap2 package [71] in normalizing the results for the ML systems is accounted for by multiplying these distance dependent quantities with bulk lattice parameters of  $\text{A}_3\text{SnO}$  compounds and dividing by 1.5 (i.e., number of layers that can be accommodated along [001] direction of the conventional anti-perovskite UC) [104]. It is clear from Fig. 8 that  $S$ ,  $\sigma_e/\tau$ , and  $\kappa_e/\tau$  for the  $\text{Ca}_3\text{SnO-ML}$  and  $\text{Sr}_3\text{SnO-ML}$  have increased near the Fermi level compared to the values computed for their bulk counterparts. As a result, both these systems show large values of the thermoelectric figure of merit  $ZT$  as high as  $\sim 0.8$  at 300 K, which is due to the Rashba-type spitting in the valence band maximum and directly linked with the enhancement of  $S$  [69]. On the contrary, all the relevant transport parameters and  $ZT$  values for  $\text{Ba}_3\text{SnO-ML}$  system show insignificant change near the Fermi level compared to its bulk counterpart. Since bulk  $\text{Ba}_3\text{SnO}$  is stable in the cubic phase above 150 K [46], we should point out here that the thermoelectric properties for this system at low temperature may be influenced by structural changes. However, the larger band overlap in  $\text{Ba}_3\text{SnO}$  would still restrict the enhancement in thermoelectric response at the 2D limit.

Although our results clearly show enhanced thermoelectric

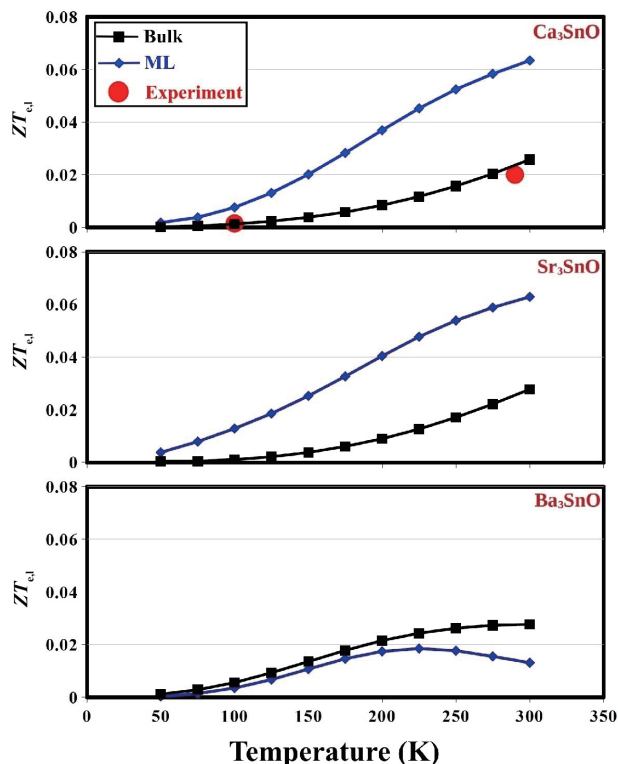
performance for  $\text{Ca}_3\text{SnO-ML}$  and  $\text{Sr}_3\text{SnO-ML}$  compared to their bulk counterparts, it is important to note that the  $\sigma_e/\tau$  and  $\kappa_e/\tau$  presented in Fig. 8 are computed within the constant relaxation time approximation ( $\tau = 10^{-14}$  s used in the present work) [71]. On the other hand, the Seebeck coefficient and figure of merit are independent of the relaxation time (typically in the range of  $\tau = 10^{-14}$  to  $10^{-15}$  [45] for bulk UC of  $\text{A}_3\text{SnO}$ ). However, the former quantity depends both on the electronic and lattice thermal conductivity for a given temperature  $T$  according to the relation

$$ZT_{e,l} = \frac{\sigma S^2 T}{\kappa_e + \kappa_l} \quad (4)$$

Thus, for a realistic evaluation of the thermoelectric properties the contribution of lattice thermal conductivity in the figure of merit for the bulk UC and ML systems studied in the present study is necessary. However, the evolution of the scattering process, lattice thermal conductivity, and relaxation times are beyond the scope of the present study since we are interested in the impact of lifting inversion symmetry on the electronic properties on going from bulk to 2D limit of the  $\text{A}_3\text{SnO}$  anti-perovskites. Nevertheless, it is crucial to provide solid evidence for the studied 2D systems to provide a rationale for motivating future experimental research efforts. To this end, we present the calculated values of  $ZT_{e,l}$  in the temperature range of 50 to 300 K for both bulk UC and ML systems by assuming a carrier concentration of  $1.5 \times 10^{-19}$  and a lattice thermal conductivity of  $\kappa_l = 2.0$  W/(m·K) that are based on the values reported in recent experiments for bulk  $\text{Ca}_3\text{SnO}$  and  $\text{Sr}_3\text{SnO}$  [43–45]. The values presented in Fig. 9 show that the calculated  $ZT_{e,l}$  for bulk  $\text{Ca}_3\text{SnO}$  are in excellent agreement with the experimental figure of merit at 300 K reported in an earlier study [43]. For both  $\text{Ca}_3\text{SnO-ML}$  and  $\text{Sr}_3\text{SnO-ML}$ , the results presented in Fig. 9 suggest that the thermoelectric performance can be enhanced by as much as  $\sim 150\%$  in the 2D system for the same level of carrier concentration as already observed in undoped bulk  $\text{A}_3\text{SnO}$  [43, 44]. This shows that the emergence of a small band gap and Rashba-type splitting in 2D MLs derived from  $\text{A}_3\text{SnO}$  anti-perovskites can further enhance their technological importance. In addition to enhancement in the thermoelectric



**Figure 8** Comparison of the variation in  $S$ ,  $\sigma_e/\tau$ ,  $\kappa_e/\tau$ , and thermometric figure of merit ( $ZT$ ) with chemical potential for bulk UC (black solid lines) and ML (blue solid lines) of (a)  $\text{Ca}_3\text{SnO}$ , (b)  $\text{Sr}_3\text{SnO}$ , and (c)  $\text{Ba}_3\text{SnO}$  at 300 K. All results have been computed using the mBJ potential functional.



**Figure 9**  $ZT_{e1}$  vs. temperature for bulk UC (black solid lines) and ML (blue solid lines) of the  $A_3SnO$  systems in the temperature range of 50 to 300 K. The experimental data for bulk  $Ca_3SnO$  reported in Ref. [43] are shown as large red circles.

properties, the electronic structure of non-magnetic  $Ca_3SnO$ -ML and  $Sr_3SnO$ -ML also indicates the potential of these 2D systems for spintronic applications where the Rashba-type splitting introduced by the SOC can be manipulated with the application of an external electric field [99].

#### 4 Conclusions

We have performed DFT calculations using reliable meta-GGA functionals to examine the impact of inversion asymmetry on the structural, thermodynamic, dynamical, mechanical, electronic, and thermoelectric properties of 1 UC thick MLs of the  $A_3SnO$  ( $A = Ca, Sr, \text{ and } Ba$ ) anti-perovskite. Our results clearly show the better performance of the SCAN + rVV10 in comparison with other exchange-correlation functionals by confirming the reproducibility of the experimental data for bulk UCs of relevant chemical systems with small average absolute deviation in the ground state structural and energetic properties obtained with this dispersion-corrected meta-GGA functional. Compared to GGA-PBE results, in particular, the SCAN + rVV10 calculations show that the stability region for the three  $A_3SnO$  compounds restricting the formation of  $CaO$ ,  $SrO$ , and  $BaO$  binary competing phases have almost similar values of atomic chemical potential of Sn atom that yield stable formation of bulk systems, in agreement with experimental observation. In addition, the mechanical and dynamical properties predicted by SCAN + rVV10 functional for the bulk UCs of  $A_3SnO$  anti-perovskite show good accord with experiment. For obtaining a reliable picture of the electronic properties of bulk UCs of  $A_3SnO$  ( $A = Ca, Sr, \text{ and } Ba$ ) compounds, the electronic band structure, density of states, and transport properties are computed using the non-local mBJ potential functional. Comparison of the DOS and ELF computed for the bulk UCs of  $A_3SnO$  show that the decreasing metallic interaction among Sn atoms in the A-Sn layer. The increasing band overlap on going from  $Ca_3SnO$  to  $Ba_3SnO$  is responsible for

closing of the band gap along  $\Gamma$ -X direction of the Brillouin zone. Since opening of the band gap at the 2D limit can be realized by removing the inversion symmetry at the 2D limit, we have also examined the structural, thermodynamic, dynamical and mechanical properties of  $A_3SnO$ -MLs by employing the SCAN + rVV10. Comparison of the calculated surface energies, phonon band structure, cohesive energies and mechanical properties with the values reported for silicene, stanene, SnSe-ML, and  $SrTiO_3$ -ML confirm the stability of the  $A_3SnO$ -MLs. The electronic properties computed using local mBJ potential functional indicate that Rashba-type splitting in the electronic structure emerges in  $Ca_3SnO$ -ML and  $Sr_3SnO$ -ML owing to strong spin-orbit coupling and inversion asymmetry in these systems. On the other hand, the  $Ba_3SnO$ -ML exhibits Dirac cone at the  $\Gamma$  symmetry point due to the dominance of band overlap in its electronic structure compared to spin-orbit coupling. By comparing the electronic transport properties of bulk and monolayer  $A_3SnO$  systems, we illustrate that the Rashba-type splitting caused by the presence of inversion asymmetry is responsible for significantly enhancing the thermoelectric properties in 2D anti-perovskites  $Ca_3SnO$ -ML and  $Sr_3SnO$ -ML.

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