Opening Up New Research Lines in Lewis Acid/Base Catalysis

Berionni, Guillaume; Hu, Lei; Ben Saida, Ali; Chardon, Aurélien; Gama, Mathieu; OSI, Arnaud; Mahaut, Damien; ANTOGNINI SILVA, Xavier

Published in:
Chimie nouvelle

Publication date:
2018

Document Version
Publisher's PDF, also known as Version of record

Link to publication

Citation for published version (HARVARD):

General rights
Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

• Users may download and print one copy of any publication from the public portal for the purpose of private study or research.
• You may not further distribute the material or use it for any profit-making activity or commercial gain.
• You may freely distribute the URL identifying the publication in the public portal?

Take down policy
If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Download date: 02. Aug. 2019
Lewis acids and bases play an important role in modern chemistry and are extensively used in academic and industrial laboratories as catalysts, ligands, initiators, scavengers, etc. [1]. In general, Lewis acids and Lewis bases combine and undergo neutralization by forming a Lewis adduct (Figure 1a) [2]. Steric bulk, however, prevents the formation of strong bonds between sterically hindered Lewis acids and bases (Figure 1b) [3].

Figure 1: a) Formation of a Lewis acid-base adduct between ammonia and boron trifluoride; b) Schematic representation of an intermolecular frustrated Lewis pair.
The resulting frustrated Lewis pairs (FLPs) with unquenched Lewis acid/base couples feature an unprecedented potential for cooperative activation of small molecules (H₂, CO₂, SO₂, CO, N₂O) [3].

Fascinated by the latest advances in the field of frustrated Lewis pairs catalysis [4], and with a longstanding interest in kinetic and mechanistic studies, we are interested in examining the fundamental chemistry of activation of small molecules (and of C-H bonds) by new types of Lewis acids and bases [5].

As a rational starting point, we are taking our inspiration from the seminal work of Wittig and Tochtermann on the reactivities of carbon and phosphorus-centred Lewis bases in combination with triphenylborane BPh₃ (Figure 3) [6, 7]. As the C–B and P–B bonds of the Lewis adducts of Ph₃C⁺ and Ph₃P with the sterically hindered BPh₃ Lewis acids are very weak, dissociation and subsequent reactions with a variety of substrates (ethers, alkenes [6] and 1,2-dehydrobenzene (benzyne) [7]) have been observed.

In sharp contrast, triphenylborane is completely associated with the ring strained triptycyl anion (Figure 4), which is more compact and basic than the trityl anion, and the resulting Lewis adduct is even air and water stable [8]. Triptycyl-lithium is also spontaneously carbonylated by CO₂ at low temperatures [9].

In view of the intense current interest in the development of sterically hindered Lewis bases [10], we are re-investigating the potential of the triptycyl anion and of the structurally related phospha [11] and aza-triptycenes [12] Lewis bases (Figure 5) as constituent of frustrated Lewis pairs in our research group.

A deep understanding of their reactivities will be obtained by performing NMR and spectrophotometrical binding titrations experiments with a series of Lewis acids of variable sizes and strengths (BF₃, BPh₃, B(C₆F₅)₃). Comparison of their stereochemical properties (Tolman cone angle θ and electronic parameter [13], pKₐ, Lewis basicity...) with those of classical and hindered phosphines will shed light on the potential applications of these triptycene-derived Lewis bases in FLPs catalysis.

These projects are synthetically challenging and expose our graduate and undergraduate students...
Opening up new research lines in Lewis acid/base catalysis

Figure 3: Early examples of reversibly formed Lewis adducts reacting with the π-bonds of 1,3-butadiene (top) and benzene (bottom) [6, 7].

Figure 4: Irreversible association of triptyc-9-yl lithium with CO₂ and BPh₃ [8, 9].

Figure 5: Examples of understudied carbon, nitrogen, and phosphorus Lewis bases.

to a variety of advanced synthetic, spectroscopic, photophysical and computational techniques. For these purposes, we have been developing collaborations with our skilled colleagues in computational chemistry, crystallography, materials and organic chemistry at the University of Namur, and are open to broad scientific discussions with anyone interested in these research areas.
Acknowledgments

We acknowledge the University of Namur, the Namur Institute of Structured Matter (NISM) and the FNRS (2018-2020 MIS grant for GB) for generous funding. We thank N. Tumanov and J. Wouters for expertise in crystallography and B. Champagne for expertise in computational chemistry. We thank the Unité de Chimie Organique team (Prof. S. Vincent, Prof. S. Lanners) for useful discussions.

Author biography

Prof. Guillaume Berionni received his PhD from the University of Versailles (Institut Lavoisier de Versailles) in 2010 for his research on physical organic and supramolecular chemistry under the supervision of Prof. F. Terrier and R. Goumont. In 2010 he moved to the Ludwig Maximilian University of Munich as a Humboldt postdoctoral fellow under the guidance of Prof. Herbert Mayr and Prof. Paul Knochel. His postdoctoral and subsequent independent research on organometallic and organoboron chemistry has been supported by the Alexander von Humbolt and the DAAD foundations.

Since end 2017 he is Professor of organic chemistry at the University of Namur (Belgium) where he is managing a research team of 3 PhD students, a post-doctoral fellow, a technician and several master students. He is also actively involved in teaching, especially at the master’s level. In 2018 he received an incentive grant for scientific research (MIS grant 2018-2020) from the FNRS.

References